Asymmetric Catalysis

Catalytic Asymmetric Intramolecular Michael Reaction of Aldehydes**

Maria T. Hechavarria Fonseca and Benjamin List*

Dedicated to Professor Johann Mulzer on the occasion of his 60th birthday

In the last few years the profound efficiency and applicability of asymmetric aminocatalysis in organic synthesis has been firmly established.^[1] After discovering the first direct asymmetric enamine catalytic intermolecular aldol, [2] Mannich, [3] Michael,^[4] and α-amination reactions,^[5] our research group became interested in using this powerful organocatalytic strategy in the design of novel intramolecular reactions in which aldehydes are used as nucleophiles. These efforts led to the first catalytic asymmetric intramolecular enolexo aldolization [Eq. (1)]^[6] and α -alkylation reactions [Eq. (2); Bn = benzyl] of aldehydes.^[7] In this context we realized that although there are a number of reports on elegant catalytic enantioselective intermolecular Michael reactions,[8] intramolecular catalytic asymmetric Michael reactions of aldehydes are [Eq. (3);EWG = electron-withdrawing group]. [9] We felt that such a process would be of great value, particularly if included in strategic reaction sequences, such as the Robinson annulation. Herein we report an efficient enantioselective cyclization of formyl enones in a process that constitutes the first catalytic asymmetric intramolecular Michael reaction of aldehydes.

Remarkable progress has been made recently in the asymmetric catalysis of intermolecular Michael reactions. Both highly effective catalytic enantioselective conjugate addition reactions of stabilized and nonstabilized carbanions^[10] and indirect Mukaivama-type intermolecular Michael reactions have been developed.^[11] In addition to these metalmediated processes, highly enantioselective organocatalytic Michael reactions, [12] including variants based on iminium and enamine catalysis,[13] have been reported. However, intramolecular catalytic asymmetric Michael reactions are extremely rare. [14] We were particularly attracted to the asymmetric aminocatalysis of intramolecular Michael reactions of formyl enones. The ketoaldehydes produced should be good substrates for a subsequent aldol condensation in an overall intramolecular Robinson annulation to yield bicyclic enones of potential use in natural product synthesis. Further-

more, such aminocatalytic Michael cyclizations are mechanistically interesting as they may proceed by either enamine or iminium catalysis, or both simultaneously.

As a model reaction we studied the amine-catalyzed Michael cyclization of formyl enone 1a to give ketoaldehyde 2a [Eq. (4)]. This reaction was catalyzed by (S)-proline (3; 10 mol %, room temperature, DMF (N,N-dimethylformamide), 3 days), but as expected, both the diastereoselectivity and enantioselectivity were low. Next, we studied the commercial imidazolidinone catalysts 4 and 5 of MacMillan and co-workers.^[15] Although these catalysts have not been used previously in enamine catalysis, [16] we found them to catalyze the cyclization reaction effectively. Interestingly, whereas the more reactive catalyst 4 provided the product in very high yield (99%) but with only 39% ee, catalyst 5 gave the product in the same yield and with an excellent 97% ee. The enantioselectivity and diastereoselectivity were determined by using an assay based on a Horner-Wadsworth-Emmons (HWE) reaction that we had developed earlier for the HPLC analysis of nonchromogenic or sensitive aldehydes.[6]

Fortunately, our results proved quite general; various other formyl enones reacted to give the desired products in excellent yields and with high *ee* values [Eq. (5), Table 1].^[17] Thus, both aromatic enones (**1a** and **1b**) and aliphatic enones (**1c** and **1d**) could be used with excellent efficiency and enantioselectivity. Remarkably, even an enal could be

^[*] Dr. M. T. Hechavarria Fonseca, Prof. Dr. B. List Max-Planck-Institut für Kohlenforschung Kaiser-Wilhelm-Platz 1, 45470 Mülheim an der Ruhr (Germany) Fax: (+49) 208-306-2999 E-mail: list@mpi-muelheim.mpg.de

^[**] We thank Linh Hoang for early experiments, Degussa for donating chemicals, and the excellent GC, HPLC, and X-ray departments of the Max-Planck-Institut für Kohlenforschung.

 Table 1:
 Catalytic asymmetric intramolecular Michael reaction of aldehydes.

Starting material ^[a]	Yield [%]	Product	ee [%]	anti/syr
O O Ia	99	H 2a	97	24:1
0 0 1b	99	H 2b	97	17:1
H 1c	99	H 2c	93	20:1
H 1d	99	H 2d	94	21:1
O O H	85	H P	80	49:1
Ts N	90	H Ts N 2f	93	8:1

[a] Ts = p-toluenesulfonyl.

employed as the Michael acceptor to furnish a dialdehyde $(1e\rightarrow 2e)$, although in somewhat reduced yield and enantioselectivity. To our delight, we were able to extend our asymmetric intramolecular Michael reaction to the synthesis of heterocyclic products. Thus, the reaction of aldehyde 1f gave the corresponding pyrrolidine 2f with excellent enantioselectivity and in good yield.

Having established an efficient process for the synthesis of ketoaldehydes **2**, we next investigated their application in two tandem sequences. Thus, the treatment of formyl enone **1c** with catalyst **5** and subjection of the product **2c** to an in situ intramolecular aldolization or HWE reaction with dimethyl (2-oxopropyl)phosphonate (**6**) gave enones **7** and **8**, respectively, in high yields and enantiomeric purity [Eq. (6); DMAP = 4-dimethylaminopyridine, Ms = methanesulfonyl]. Hydrindenone **7** has been described previously as the racemate, but this is the first enantioselective synthesis of **7**.^[18] Both the bicyclic enone **7** and diketone **8** should be useful

additions to the chiral pool and valuable precursors in natural product synthesis.

We assume the Michael reaction described herein to proceed through an enamine mechanism. However, the observation that only enones react to give the desired products with very high enantioselectivity may be interpreted as evidence for an intriguing dual-activation mechanism involving both enamine and iminium catalysis. Furthermore, an inverse-electron-demand hetero-Diels-Alder mechanism, reminiscent of the intramolecular enamine/enal cycloaddition of Schreiber et al.^[19] and the organocatalytic hetero-Diels-Alder reaction of Juhl and Jørgensen,^[20] should be considered.

In summary, we have disclosed the first catalytic asymmetric intramolecular Michael reaction of aldehydes. Our process furnishes cyclic ketoaldehydes in excellent yields, with high enantioselectivities, and under very mild and convenient reaction conditions. This reaction is therefore expected to find use in organic synthesis. Further studies towards expanding the scope of this reaction and developing a more detailed mechanistic understanding are well underway in our laboratory.

Experimental Section

Typical procedure: Ketoaldehyde 1c (123 mg, 0.8 mmol) was dissolved in dry THF (8 mL) and treated with 5 (20.4 mg, 0.08 mmol, 10%). The resulting mixture was stirred under argon at room temperature until the starting material had disappeared (15 h). The solution was then concentrated and filtered through silica. The product 2c was dissolved in methanol (4 mL), cooled to 0°C, and treated with NaOH (2N, 4 mL).[21] After 5 min, TLC analysis indicated completion of the reaction. The mixture was poured into saturated aqueous ammonium chloride solution, the resulting mixture was extracted with ether, and the extracts were dried, filtered, and concentrated to give the crude aldol product. This material was dissolved in CH₂Cl₂ (5.2 mL) and stirred with methanesulfonvl chloride (0.16 mL), DMAP (35.6 mg), and triethylamine (0.39 mL) under argon for 30 min. After an aqueous workup and purification by chromatography on silica gel (20% ether/pentane), enone 7 was obtained as a colorless oil (95.2 mg, 0.7 mmol, 88 %).[22]

Received: May 7, 2004 [Z460578] Published Online: July 7, 2004

Keywords: asymmetric catalysis · cyclization · enamines · Michael reactions · organocatalysis

Zuschriften

- For selected reviews, see: a) B. List, Synlett 2001, 1675-1686;
 b) H. Gröger, J. Wilken, Angew. Chem. 2001, 113, 545-548;
 Angew. Chem. Int. Ed. 2001, 40, 529-532;
 c) P. I. Dalko, L. Moisan, Angew. Chem. 2001, 113, 3840-3864;
 Angew. Chem. 1113, 3840-3864;
 Angew. Chem. Int. Ed. 2001, 40, 3726-3748;
 d) S. Borman, Chem. Eng. News 2002, 80(50), 35;
 e) E. R. Jarvo, S. J. Miller, Tetrahedron 2002, 58, 2481-2495;
 f) B. List, Tetrahedron 2002, 58, 5572-5590;
 g) M. Movassaghi, E. N. Jacobsen, Science 2002, 298, 1904-1905;
 h) B. List, Acc. Chem. Res. 2004, in press.
- [2] a) B. List, R. A. Lerner, C. F. Barbas III, J. Am. Chem. Soc. 2000, 122, 2395–2396; b) W. Notz, B. List, J. Am. Chem. Soc. 2000, 122, 7386–7387; c) B. List, P. Pojarliev, C. Castello, Org. Lett. 2001, 3, 573–575.
- [3] B. List, J. Am. Chem. Soc. 2000, 122, 9336–9337; b) B. List, P. Pojarliev, W. T. Biller, H. J. Martin, J. Am. Chem. Soc. 2002, 124, 827–833; c) P. Pojarliev, W. T. Biller, H. J. Martin, B. List, Synlett 2003, 1903–1904.
- [4] B. List, P. Pojarliev, H. J. Martin, Org. Lett. 2001, 3, 2423-2425.
- [5] B. List, J. Am. Chem. Soc. 2002, 124, 5656-5657.
- [6] C. Pidathala, L. Hoang, N. Vignola, B. List, Angew. Chem. 2003, 115, 2891–2894; Angew. Chem. Int. Ed. 2003, 42, 2785–2788.
- [7] N. Vignola, B. List, J. Am. Chem. Soc. 2004, 126, 450–451.
- [8] For important recent reviews, see: a) N. Krause, A. Hoffmann-Röder, Synthesis 2001, 171–196; b) J. Christoffers in Encyclopedia of Catalysis, Vol. 5 (Ed. I. Horvath), Wiley, New York, 2002, pp. 99–118; c) J. Christoffers, A. Baro, Angew. Chem. 2003, 115, 1726–1728; Angew. Chem. Int. Ed. 2003, 42, 1688–1690.
- [9] For non-asymmetric catalytic intramolecular Michael reactions of aldehydes via enamine intermediates, see, for example: a) S. D. Burke, C. W. Murtiashaw, J. O. Saunders, M. S. Dike, J. Am. Chem. Soc. 1982, 104, 872-874; b) H. Hagiwara, N. Komatsubara, H. Ono, T. Okabe, T. Hoshi, T. Suzuki, M. Ando, M. Kato, J. Chem. Soc. Perkin Trans. 1 2001, 316-322; c) for an elegant cobalt-catalyzed Michael cycloreduction, see: T.-G. Baik, A. L. Luis, L.-C. Wang, M. J. Krische, J. Am. Chem. Soc. 2001, 123, 5112-5113; d) for the base-catalyzed intramolecular Michael reaction/aldol condensation of aldehydes in the synthesis of hydrindenones, see: G. Stork, C. S. Shiner, J. D. Winkler, J. Am. Chem. Soc. 1982, 104, 310-312.
- [10] See, for example: a) B. L. Feringa, Acc. Chem. Res. 2000, 33, 346–353; b) M. T. Reetz, A. Gosberg, D. Moulin, Tetrahedron Lett. 2002, 43, 1189–1191; c) Y. Takaya, M. Ogasawara, T. Hayashi, J. Am. Chem. Soc. 1998, 120, 5579–5580; d) Y. S. Kim, S. Matsunaga, J. Das, A. Sekine, T. Ohshima, M. Shibasaki, J. Am. Chem. Soc. 2000, 122, 6506–6507.
- [11] For a catalytic asymmetric Mukaiyama–Michael reaction, see: D. A. Evans, K. A. Scheidt, J. N. Johnston, M. C. Willis, J. Am. Chem. Soc. 2001, 123, 4480–4491.
- [12] For selected Brønsted base catalyzed intermolecular asymmetric Michael reactions, see: a) K. Hermann, H. Wynberg, J. Org. Chem. 1979, 44, 2238–2244; b) T. Ishikawa, Y. Araki, T. Kumamoto, H. Seki, K. Fukuda, T. Isobe, Chem. Commun. 2001, 245–246; for selected intermolecular asymmetric Michael reactions under phase-transfer catalysis, see: c) R. S. E. Conn, A. V. Lovell, S. Karady, L. M. Weinstock, J. Org. Chem. 1986, 51, 4710–4711; d) E. J. Corey, F.-Y. Zang, Org. Lett. 2000, 2, 4257–4259.
- [13] For selected iminium catalytic intermolecular asymmetric Michael reactions, see: a) M. Yamaguchi, T. Shiraishi, M. Hirama, J. Org. Chem. 1996, 61, 3520; b) A. Kawara, T. Taguchi, Tetrahedron Lett. 1994, 35, 8805; c) S. Hanessian, V. Pham, Org. Lett. 2000, 2, 2975; d) N. A. Paras, D. W. C. MacMillan, J. Am. Chem. Soc. 2001, 123, 4370; e) N. A. Paras, D. W. C. MacMillan, J. Am. Chem. Soc. 2002, 124, 7894-7895; f) S. P. Brown, N. C. Goodwin, D. W. C. MacMillan, J. Am. Chem. Soc. 2003, 125,

- 1192–1194; g) N. Halland, R. G. Hazell, K. A. Jørgensen, *J. Org. Chem.* **2002**, 67, 8331–8338; h) N. Halland, P. S. Aburel, K. A. Jørgensen, *Angew. Chem.* **2003**, 115, 685–689; *Angew. Chem. Int. Ed.* **2003**, 42, 661–665; i) N. Halland, T. Hansen, K. A. Jørgensen, *Angew. Chem.* **2003**, 115, 5105–5107; *Angew. Chem. Int. Ed.* **2003**, 42, 4955–4957; for enamine catalytic intermolecular asymmetric Michael reactions, see: reference [4]; j) J. M. Betancort, K. Sakthivel, R. Thayumanavan, C. F. Barbas III, *Tetrahedron Lett.* **2001**, 42, 4441–4444; k) D. Enders, A. Seki, *Synlett* **2002**, 26–28; l) P. Melchiorre, K. A. Jørgensen, *J. Org. Chem.* **2003**, 68, 4151–4157; m) O. Andrey, A. Alexakis, G. Bernardinelli, *Org. Lett.* **2003**, 5, 2559–2561.
- [14] a) A. P. Kozikowski, B. B. Mugrage, J. Org. Chem. 1989, 54, 2275-2277; b) Y. Hirai, T. Terada, T. Yamazaki, T. Momose, J. Chem. Soc. Perkin Trans. 1 1992, 509-516; c) M. G. Banwell, D. A. S. Beck, J. A. Smith, Org. Biomol. Chem. 2004, 2, 157-159; for auxiliary-mediated asymmetric intramolecular Michael reactions, see, for example: d) D. A. Evans, M. T. Bilodeau, T. C. Somers, J. Clardy, D. Cherry, Y. Kato, J. Org. Chem. 1991, 56, 5750-5752; e) F. Dumas, J. d'Angelo, Tetrahedron: Asymmetry 1990, 1, 167.
- [15] a) K. A. Ahrendt, C. J. Borts, D. W. C. MacMillan, J. Am. Chem. Soc. 2000, 122, 4243-4244; b) J. F. Austin, D. W. C. MacMillan, J. Am. Chem. Soc. 2002, 124, 1172-1173.
- [16] During the preparation of this manuscript, MacMillan and coworkers disclosed an enamine-catalytic α-chlorination reaction of aldehydes with catalyst 5: M. P. Brochu, S. P. Brown, D. W. C. MacMillan, J. Am. Chem. Soc. 2004, 126, 4108–4109.
- [17] Other Michael acceptors were either unreactive (X = CH₂, EWG = CO₂Et or COSEt [Eq. (3)]) or less stereoselective (EWG = NO₂, 96% yield, anti/syn = 26:1, 51% ee.)
- [18] S. W. Baldwin, H. R. Blomquist, Jr., J. Am. Chem. Soc. 1982, 104, 4990 – 4992.
- [19] S. L. Schreiber, H. V. Meyers, K. B. Wiberg, J. Am. Chem. Soc. 1986, 108, 8274–8277.
- [20] K. Juhl, K. A. Jørgensen, Angew. Chem. 2003, 115, 1498-1501; Angew. Chem. Int. Ed. 2003, 42, 1536-1539.
- [21] J. H. Hutchinson, T. Money, Can. J. Chem. 1987, 65, 1-6.
- [22] The absolute and relative configuration of enone 7 was determined after hydrogenation to the known saturated derivative; see: A. R. Krawczyk, J. B. Jones, *J. Org. Chem.* 1989, 54, 1795–1801.